

TWO NEW TRICOTHECENES PRODUCED BY *FUSARIUM* sp.

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Abstract—Two new 12,13-epoxytrichothecenes have been isolated from the culture filtrate of *Fusarium* sp. strain K-5036 and characterized as 4 β ,15-diacetoxy-12,13-epoxytrichothec-9-ene-3 α ,7 α -diol and 4 β ,15-diacetoxy-12,13-epoxytrichothec-9-ene-3 α ,7 α ,8 α -triol. In addition, the fungus produced diacetoxyscirpenol, nivalenol diacetate and neosolaniol.

INTRODUCTION

The naturally occurring fungal sesquiterpenoids, having the trichothecene nucleus, have toxic effects on animals, plants and micro-organisms [1]. This class of toxic metabolites was isolated from various species of *Trichothecium*, *Trichoderma*, *Myrothecium*, *Cephalosporium*, *Fusarium* and *Stachybotrys*; and so far 13 trichothecenes have been isolated from species of *Fusarium* [2,3].

Mycotoxicological studies on the metabolites of *Fusarium* spp. led to the discovery of two new trichothecenes. The present paper describes the isolation and structures of the newly discovered toxins.

RESULTS AND DISCUSSION

Toxins were extracted from the culture filtrate of *Fusarium* sp. strain K-5036 grown on Czapek-Dox-peptone medium using charcoal adsorption and methanol elution method [4]. Extraction of methanol-chloroform soluble materials with acetone gave a yellow powder which was divided into 8 fractions (A-H) by chromatography on a Si gel column. Fractions were tested for toxicity using mice and rabbit reticulocytes [5]; toxicity was detected in fractions B-F. TLC analysis of the fractions revealed the presence of diacetoxyscirpenol (6) [6,7] and nivalenol diacetate (7) [8,9] in fraction B and neosolaniol (8) [10,11] in frac-

tion D. Further purifications of fractions C and E by chromatography on Si gel columns gave crystals of (1) and (2), respectively.

	R ¹	R ²	R ³	
			α	β
(1)	OH	OH	H	H
(2)	OH	OH	OH	H
(3)	OAc	OAc	H	H
(4)	OAc	OH	OAc	H
(5)	OAc	OAc	OAc	H
(6)	OH	H	H	H
(7)	OH	OH	O	
(8)	OH	H	OH	H
(9)	OH	OH	OAc	H

Compound (1) has an empirical formula of C₁₉H₂₆O₈ (elemental analysis) and is an isomer of (8). The IR spectrum, being similar to that of (8), indicated the presence of hydroxyl (3500 cm⁻¹) and acetyl (1720 and 1280 cm⁻¹) groups. The MS (probe) showed the base peak at *m/e* 43 corresponding to acetylation. The M⁺ did not appear but M⁺-18 (*m/e* 364) and M⁺-60 (*m/e* 322) indicating the presence of hydroxyl and acetoxy groups occurred in the high mass region.

The NMR spectrum of (1) showed signals characteristic of the naturally occurring trichothecenes such as the AB quartet at δ 3.07 and 3.17 arising from the spiroepoxy protons (H-13) and the two methyl singlets at δ 1.13 and 1.75 due to the tertiary methyl (H-14) and the allylic methyl (H-16) groups, respectively. The presence of two

Table 1. Chemical shifts (δ values) of protons in (1) and (2) and their acetates

Compound	2-H	3-H	4-H	7-H	8-H	10-H	11-H	13-H	14-H	15-H	16-H	OAc
(1)	3.70d (5)	ca 4.25 (3.5)	5.20d (3.5)	4.62dd (6.10)	ca 2.0, 2.43dd	5.50d (6)	ca 4.25	3.07d, 3.17d (4)	1.13s	4.24d, 4.42d (12)	1.75s, 2.07s, 2.15s	
(2)	3.70d (5)	ca 4.2 (3)	5.45d (3)	4.01d (5)	ca 4.5	5.65d (6)	ca 4.5	3.05d, 3.15d (4)	1.10s	4.29d, 4.50d	1.85s, 2.02s, 2.13s	
(3)	3.88d	5.17d	6.10d	5.73d	ca 2.05, 2.65dd	5.55d	4.42d	2.64d, 3.10d	0.90s	4.18d, 4.65d	1.70s, 2.05s, 2.08s, 2.11s, 2.14s	
(4)	3.90d	5.20d	5.92d	4.70d	5.50d	5.80d	4.32d	3.10d, 3.25d	1.10s	4.35d, 4.55d	1.78s, 2.08s, 2.11s, 2.14s, 2.14s	
(5)	3.90d	5.18d	6.05d	5.60d	5.85d	5.80d	4.55d	2.65d, 3.05d	0.85s	4.40d, 4.70d	1.80s, 2.04s, 2.1s, 2.14s, 2.14s, 2.19s	

Coupling constants in parentheses; s, singlet; d, doublet; dd, double doublet.

acetyl groups were revealed by two 3-proton singlets at δ 2.07 and 2.14, and the chemical shifts of the signals due to H-4 (δ 5.20) and H-15 (δ 4.33) suggested the presence of acetoxy groups at positions 4 and 15. The presence of a hydroxyl group at position 3 was also indicated by the chemical shift of the signal due to H-3 (δ 4.25) which is coupled with the signals due to H-2 (δ 3.70) and H-4 (δ 5.20). The coupling constants, $J_{2,3}$ (5 Hz) and $J_{3,4}$ (3.5 Hz), indicated the orientation of the hydroxyl group at position 3 was α and that of the acetoxy group at position 4 was β .

Spin-decoupling experiments revealed the presence of an ABX system at δ 2.05, 2.43 and 4.62. The signals were due to the partial structure $-\text{CH}_2-\text{CH}(\text{OH})-$ which could occupy position 7–8. The coupling constants of the X signal at δ 4.62 (J 6, 10 Hz) indicated a 7 α - or 8 β -hydroxyl group. On the other hand, deshielding of H-14 (δ 1.13) compared with that of (8) (δ 0.85) suggested the presence of a 7 α -oriented hydroxyl group in (1) [12].

The structure of (1) was confirmed by analysis of the NMR spectrum of its diacetate (3). Acetylation resulted in downshifts of the signals due to H-3 (0.92 ppm) and H-7 (1.11 ppm), and an upshift of the signal due to H-14 (0.23 ppm).

Compound (2), $\text{C}_{19}\text{H}_{26}\text{O}_9$, showed an IR spectrum similar to those of (1) and (8). The MS showed the M^+ at m/e 398 and the base peak at m/e 43. The NMR spectrum indicated the presence of 3 α -hydroxyl, 7 α -hydroxyl, 4 β -acetoxy and 15-acetoxy groups as in the case of (1).

The significant difference between the NMR spectra of (1) and (2) was that the latter did not show the methylene signal due to H-8. This difference and the chemical formula suggested that one hydroxyl group was present at position 8. The orientation of the hydroxyl group on C-8 was revealed by acetylation of (2) and measuring the

coupling constant $J_{7,8}$. Acetylation gave a diacetate (4) and a triacetate (5); the coupling constant of the signal due to H-8 (5 Hz) in the NMR spectrum of (4) indicated a 8 α -hydroxyl group.

Chemical proof of the structure of (2) was provided by the NaBH_4 reduction of (7) to (2). The reaction was previously reported [13] although mp of the product obtained in the previous study was different from this compound. A triacetate (9) of the parent alcohol of (2) was previously isolated from *F. scirpi* [6,14]. *Fusarium* sp. K-5036 thus produces five toxins (1, 2, 5, 7, and 8) which differ only in their substitutions at positions 7 and 8 in the trichothecene nucleus.

EXPERIMENTAL

Mp's are uncorrected. IR spectra were taken as KBr pellets. NMR spectra were measured for solutions in CDCl_3 with TMS as internal standard. Kieselgel 60 and Kieselgel G (Merck) were used for column and TLC, respectively. For PLC, 500 μm adsorbent layers were used. Spots on TLC plates were made visible by spraying with 20% H_2SO_4 and heating at 110°.

Culture of fungus. The strain of *Fusarium* sp. K-5036 used in this study was isolated from a sample of river water and kindly donated by Dr. Y. Matsuda, Koube City Institute of Environmental Hygienics. Inoculum was grown at 25° on potato-dextrose agar slant for 14 days. Conidia and mycelia of the fungus were seeded in 100 Fernbach flasks each containing 250 ml of Czapek-Dox-peptone soln [4] and incubated stationary at 25° for 14 days.

Extraction of toxins. Procedure for extraction of toxins from the culture filtrate was carried out with the method previously reported [4] with modifications. Culture filtrate (21.5 l) was mixed with activated charcoal (200 g) and kept at room temp. for 3 hr with occasional stirring. The charcoal collected by filtration was washed with H_2O and the materials adsorbed on the charcoal were eluted 2 \times with MeOH (4, 2 l). Combined MeOH eluates were evaporated to dryness and redissolved in 250 ml of hot MeOH; after removal of insoluble materials, the soln was concentrated to 100 and 500 ml of CHCl_3 was added resulting in formation of ppt. The MeOH- CHCl_3 soln was filtered and the filtrate was evaporated to dryness; the powdered material (10.5 g) was extracted with 300 ml of Me_2CO under reflux to yield Me_2CO soluble material (5.5 g).

4 β ,15-diacetoxy-12,13-epoxytrichothec-9-ene-3 α ,7 α -diol. Me₂CO extract was chromatographed on a column (4 × 80 cm) packed with Si gel and developed with C₆H₆-Me₂CO (5:2, 1500 ml; 2:1, 1500 ml; 1:1, 1000 ml) followed by Me₂CO (1000 ml) and MeOH (1000 ml). The eluate was monitored by TLC and divided into 8 fractions (A 99, B 434, C 288, D 94, E 532, F 707, G 445, H 2000 mg) according to the *R_f* values of the components. Fraction C (280 mg) was further purified by chromatography on a Si gel column (1.5 × 40 cm) with *n*-hexane-Me₂CO (1:1). The purified material which gave *R_f* 0.5 on TLC with the same system (200 mg) was crystallized from CHCl₃ and *n*-hexane as needles. The compound was recrystallized 2× from the same solvents (107 mg), mp 201–203° (Found: C, 59.25; H, 6.78. C₁₉H₂₆O₈ requires: C, 59.65; H, 6.85%); IR bands at 3500, 2950, 1720, 1380, 1280 and 1060 cm⁻¹; MS *m/e* 364, 322, 304 and 263.

4 β ,15-diacetoxy-12,13-epoxytrichothec-9-ene-3 α ,7 α ,8 α -triol. Fraction E (525 mg) was also chromatographed on a Si gel column (1.5 × 50 cm) with *n*-hexane-Me₂CO (1:1). The material giving *R_f* 0.38 on TLC with the same solvent system (380 mg) gave needles from EtOH-Et₂O-*n*-hexane. The compound was recrystallized from the same solvents (200 mg), mp 167–169° (Found: C, 56.56; H, 6.50. C₁₉H₂₆O₉ requires C, 57.28; H, 6.58%); IR bands at 3500, 2950, 1750, 1390, 1260 and 1050 cm⁻¹; MS *m/e* 398, 380, 338, 320, 277 and 261.

Acetylation of (1). (1) (30 mg), pyridine (1.5 ml) and Ac₂O (1.5 ml) was set aside at room temp. for 24 hr. The product was purified by PLC, developed with *n*-hexane-Me₂CO (2:1), and eventually crystallized from C₆H₆-*n*-hexane to give 3 α ,4 β ,7 α ,15-tetraacetoxy-12,13-epoxytrichothec-9-ene as needles (25 mg), mp 147–149°.

Acetylation of (2). (2) (25 mg) in pyridine (5 ml) and Ac₂O (5 ml) was kept at room temp. for 18 hr. The crude product was subjected to PLC with *n*-hexane-Me₂CO (2:1). Two major bands (*R_f* 0.36 and 0.27) were scraped off and eluted with Me₂CO. Solid obtained from the lower band was crystallized with C₆H₆-*n*-hexane to give 3 α ,4 β ,8 α ,15-tetraacetoxy-12,13-epoxytrichothec-9-ene-7 α -ol as needles (9.5 mg), mp 207.5–209.5°. Evaporation of eluate from the upper band gave amorphous 3 α ,4 β ,7 α ,8 α ,15-pentaacetoxy-12,13-epoxytrichothec-9-ene (12 mg).

NaBH₄ reduction of (7). (7) (50 mg) in MeOH (2 ml) was treated at 0° with NaBH₄ (35 mg) in MeOH (1 ml) and kept at room temp. for 2 hr. Addition of H₂O and HOAc (to pH 5) was followed by evaporation of MeOH and continuous

extraction with CHCl₃ for 2 hr. The extract was then subjected to PLC with *n*-hexane-Me₂CO (1:1). The band corresponding to (2) (*R_f* 0.24) was scraped off and eluted with Me₂CO. Evaporation of eluate gave a solid which crystallized from EtOH-Et₂O-*n*-hexane as needles. The mp, IR and NMR spectra of the crystal were identical with those of (2).

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